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Dynamics of competition model between two plants based on stoichiometry

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Abstract: The dynamics of two-plant competitive models have been widely studied, while the effect of chemical heterogeneity on competitive plants is rarely explored. In this study, a model that explicitly incorporates light and total phosphorus in the system is formulated to characterize the impacts of limited carbon and phosphorus on the dynamics of the two-plant competition system. The dissipativity, existence and stability of boundary equilibria and coexistence equilibrium are proved, when the two plants compete for light equally. Our simulations indicate that, with equal competition for light ($b_{12} = b_{21}$) and a fixed total phosphorus in the system (T), plants can coexist with moderate light intensity (K). A higher K tends to favor the plant with a lower phosphorus loss rate (d_1 vs d_2). When K is held constant, a moderate level of T leads to the dominance of the plant with a lower phosphorus loss rate (d_1 vs d_2). At high T levels, both plants can coexist. Moreover, our numerical analysis also shows that, when the competition for light is not equal, the low level of total phosphorus in the system may lead the model to be unstable and have more types of bistability compared with the two-dimensional Lotka-Volterra competition model.

Keywords: stoichiometric system; competition model; nutrient limitation; bistability; bifurcation

1. Introduction

The two-dimensional Lotka-Volterra competition model has been widely studied [1–4]. Murray discussed a simple model on competition between two species and explains the biological conclusion implied by the competition exclusion principle, namely, when two species compete for the same limited resources, one of the species usually becomes extinct [1]. Song et al. considered the Lotka-Volterra competitive system with two delays and studied the dynamical properties of the system [2]. Jin et al. considered a nonautonomous two-dimensional competitive Lotka-Volterra system with impulse and gave an extension of the principle of competition exclusion [3]. Zhang and Chen studied the effects of linear and nonlinear diffusion of the competitive Lotka-Volterra model [4]. Hsu and Zhao analyzed the stability of a monotone dynamical system and obtained the global dynamics of the Lotka-Volterra

two-species competition model with seasonal succession [5]. Wang and Zhao investigated two free boundary problems of a Lotka-Volterra type competition model in a one-dimension space to understand the asymptotic behavior of the two competing species spreading through a free boundary [6]. Liu and Fan proposed a new definition of permanence for stochastic population models, which overcame some limitations and deficiency of the existing ones [7]. Ren and Liu considered the two-species chemotaxis system with Lotka-Volterra competitive kinetics in a bounded domain with smooth boundary [8].

In fact, all organisms are composed of multiple chemical elements such as carbon, nitrogen and phosphorus (Loladze et al. [9]). Ecological stoichiometry is the balance of multiple chemical substances in ecological interactions and processes, or the study of such balances (Sterner and Elser [10]). From the stoichiometric point of view [9, 10], both food quantity and quality need to be explicitly modeled in producer-consumer interactions, as it has been observed that the plant quality and quantity can greatly affect the growth rate of herbivorous grazers [11–17]. Loladze et al. applied stoichiometric principles to construct a two-dimensional Lotka-Volterra type model (LKE model) that links energy flow with element cycling and explains the Rosenzweig's paradox of enrichment [9]. Wang et al. proposed a stoichiometric producer-grazer model, which extended the LKE model by taking P into account in both the prey and the media [18]. The results show that as the uptake rate of P by producer becomes infinite, LKE models become the limiting case of this model. Peace et al. formulated a model to explicitly track free P in the stoichiometric knife-edge model. The model shows that the fate of the grazer population can be very sensitive to excessive nutrient concentrations [19].

Furthermore, the growing study of ecological stoichiometry also further facilitates more insights on mechanisms of competitive systems between the two species. For example, Tilman et al. explained the mechanism of plant competition for nutrients by giving the expression of plant competition for nutrients [20]. Ji and Wang incorporated stoichiometry into a competition model in chemostat settings [21] and considered two species competing for a single nutrient, and investigated how stoichiometry, dilution rate and concentration of phosphorus input affected the result of competition between algae species. However, for the system with both interspecific and interspecific competition in [1], the effect of nutrients on the growth of both species was unclear and an ecological stoichiometric model was essential.

In this study, by considering the ecological stoichiometry of nutrient elements, we formulate a new stoichiometric competition model to investigate the dynamics among two plants in section two. Here, the phosphorus in the system is divided into three pools: Phosphorus in the first plant, phosphorus in the second plant and free phosphorus. We analyze the positive invariant region of the model and derive the existence and stability conditions of the equilibria. In section three, we first study the stability of boundary equilibria in Case 1, then, we provide the solution curves of the model and verify the results of theoretical analysis. Finally, we study the bifurcation diagram of the system about K and T, and the bistability of the system with interspecific competition is employed to investigate how total phosphorus in the system and light intensity affect the competition outcomes. Section four concludes and discusses the implications of our mathematical findings.

2. Modeling the competition between two plants

In this section, we present how we formulate the mathematical model for the competition of two plants. Next, we derive the equilibria of the model and analyze the stability.

2.1. Model formulation

In this work, we integrate stoichiometry into the competition model of two plants. We start with a two-species Lotka-Volterra competition model [1]:

$$\begin{cases} \frac{dx_1}{dt} = r_1 x_1 \left(1 - \frac{x_1}{K_1} - b_{12} \frac{x_2}{K_1} \right), \\ \frac{dx_2}{dt} = r_2 x_2 \left(1 - \frac{x_2}{K_2} - b_{21} \frac{x_1}{K_2} \right). \end{cases}$$
(2.1)

In model (2.1), x_i (i = 1, 2) describe the biomasses of the plant measured in terms of carbon. r_1 , r_2 , K_1 , K_2 , b_{12} and b_{21} are all positive constants, r_i (i = 1, 2) are the linear birth rates and K_i (i = 1, 2) are the carrying capacities. We use b_{12} and b_{21} to measure the competitive effect of x_1 on x_2 and x_2 on x_1 , respectively. Model (2.1) only considers the energy flow in a species. However, element cycling cannot be ignored in a competition model. Hence, it is realistic to take into account the stoichiometric competition model.

We assume that plants are composed of two main elements, carbon and phosphorus, and we consider two plants, plant one and plant two. Let T represent the total phosphorus in the system. p_i (i = 1, 2) denotes the total phosphorus in plant *i*, and its unit is (mg P)/L. b_{12} and b_{21} are the competition coefficients between two plants. Following the main assumptions used in the LKE model [9], we assume

(i) The total phosphorus in the system (T) in the entire system is fixed.

(ii) Phosphorus to carbon ratio (P:C) varies in plants, but it never falls below a minimum ratio for plant one and plant two, denoted by q_1 and q_2 , respectively.

Recall that P:C in the plants should be at least q_i (i = 1, 2) ((mg P)/(mg C)), and one obtains that the plant density cannot exceed p_i/q_i ((mg P)/L). We consider the competition of plants for carbon (light) and the restriction of phosphorus on the growth of each plant. The internal phosphorus dependent growth function follows the empirically well tested Droop form, $1 - \frac{q_i}{p_i/x_i}$ (i = 1, 2) [22]. Then, the growth is restricted by the light intensity (K) and the total phosphorus in the system ($p_1/q_1, p_2/q_2$). The model takes the form of

$$\begin{cases} \frac{dx_1}{dt} = r_1 x_1 \min\left\{1 - \frac{x_1 + b_{12} x_2}{K}, 1 - \frac{q_1}{p_1/x_1}\right\},\\ \frac{dx_2}{dt} = r_2 x_2 \min\left\{1 - \frac{b_{21} x_1 + x_2}{K}, 1 - \frac{q_2}{p_2/x_2}\right\},\\ \frac{dp_1}{dt} = g_1 (T - p_1 - p_2) x_1 - d_1 p_1,\\ \frac{dp_2}{dt} = g_2 (T - p_1 - p_2) x_2 - d_2 p_2. \end{cases}$$

$$(2.2)$$

Here, $T - p_1 - p_2$ is the free phosphorus in the environment. Let d_i (i = 1, 2) be the loss rate of phosphorus in the plants, then the phosphorus element metabolized by plants is $d_i p_i$ (i = 1, 2). Hence, the biological significance of other parameters is explained below. Parameter values estimated

by Wang et al. (2008) [18] are listed in Table 1. The replenishment rate functions $g_i(y)$ are increasing, nonnegative for nonnegative variables. In general, $g_i(y)$ is a bounded smooth function that satisfies the following assumptions:

$$g_i(y) = 0, \ g'_i(y) > 0, \ \text{and} \ g''_i(y) \le 0, \ \text{for } y \ge 0.$$
 (2.3)

Parameter	Interpretation	Value	Unit
r_1	Intrinsic growth rate of plant one	0.93	day ⁻¹
r_2	Intrinsic growth rate of plant two	1.2	day^{-1}
K	light intensity	0.25-2.0	(mg C)/L
q_1	Minimal cell quota of plant one	0.003	(mg P)/(mg C)
q_2	Minimal cell quota of plant two	0.004	(mg P)/(mg C)
Т	Total phosphorus in the system	0.025	mgP/L
c_1	Maximal phosphorus uptake rate of plant one	0.2	(mg P)/(mg C)/day
<i>c</i> ₂	Maximal phosphorus uptake rate of plant two	0.18	(mg P)/(mg C)/day
d_1	Plant one loss rate of phosphorus	0.05	day^{-1}
d_2	Plant two loss rate of phosphorus	0.06	day^{-1}
b_{12}	Influence coefficient of plant two on plant one	1.0	dimensionless
b_{21}	Influence coefficient of plant one on plant two	1.0	dimensionless
a_1	Half-saturation constant of phosphorus for plant one	0.008	(mg P)/L
a_2	Half-saturation constant of phosphorus for plant two	0.01	(mg P)/L

Table 1. The parameters of (2.2) and their values used for numerical simulations.

To prevent the orbits from entering the origin where the system is undefined, we devise a transformation to generate a new system in [15]. The transformation converts variables plant C and plant P to variables plant C and plant P:C ratio. We introduce $Q_1 = p_1/x_1$ and $Q_2 = p_2/x_2$, which are the cell quota (intracellular P:C ratio) in two plants. We set the transformation

$$\Phi: \Omega \to \Phi(\Omega), (x_1, x_2, p_1, p_2) \mapsto (x_1, x_2, Q_1 = p_1/x_1, Q_2 = p_2/x_2),$$

which converts (2.2) in Ω into the following system

$$\begin{cases} \frac{dx_1}{dt} = r_1 x_1 \min\left\{1 - \frac{x_1 + b_{12} x_2}{K}, 1 - \frac{q_1}{Q_1}\right\}, \\ \frac{dx_2}{dt} = r_2 x_2 \min\left\{1 - \frac{b_{21} x_1 + x_2}{K}, 1 - \frac{q_2}{Q_2}\right\}, \\ \frac{dQ_1}{dt} = g_1 (T - Q_1 x_1 - Q_2 x_2) - d_1 Q_1 - Q_1 r_1 \min\left\{1 - \frac{x_1 + b_{12} x_2}{K}, 1 - \frac{q_1}{Q_1}\right\}, \\ \frac{dQ_2}{dt} = g_2 (T - Q_1 x_1 - Q_2 x_2) - d_2 Q_2 - Q_2 r_2 \min\left\{1 - \frac{b_{21} x_1 + x_2}{K}, 1 - \frac{q_2}{Q_2}\right\}. \end{cases}$$
(2.4)

Here, $\Phi(\Omega) = \{(x_1, x_2, Q_1, Q_2): 0 < x_1 < \min\{K, T/q_1\}, 0 < x_2 < \min\{K, T/q_2\}, Q_1 > q_1, Q_2 > q_2, Q_1x_1 + Q_2x_2 < T\}.$

More details of mathematical analysis can be found in Appendix A.

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3. Numerical dynamics and the implications

In this section, we first analyze the stability of the two boundary equilibria in Case 1, and then we analyze the bifurcation of (A.1). Finally, we study the stability of the system with interspecific competition. The initial conditions in our simulations are chosen inside the biologically meaningful region.

3.1. Stability analysis of the boundary equilibria in Case 1

In this subsection, we analyze the stability of the two boundary equilibria in Case 1. The model (A.2) has two boundary equilibria

$$E_1 = \left(0, K, \frac{c_1 T d_2}{d_1 (c_2 K + d_2)}, \frac{c_2 T}{c_2 K + d_2}\right), \ E_2 = \left(K, 0, \frac{c_1 T}{c_1 K + d_1}, \frac{c_2 T d_1}{d_2 (c_1 K + d_1)}\right).$$

Based on the study of Theorem 2, the first two equations of (A.2) do not contain variables Q_1 and Q_2 . Here, we consider the submodel of (A.2)

$$\begin{cases} \frac{dx_1}{dt} = r_1 x_1 \left(1 - \frac{x_1 + x_2}{K} \right) = f(x_1, x_2), \\ \frac{dx_2}{dt} = r_2 x_2 \left(1 - \frac{x_1 + x_2}{K} \right) = g(x_1, x_2). \end{cases}$$
(3.1)



Figure 1. The phase diagram of the submodel (3.1).

We present the phase diagram of (3.1). In Figure 1, $r_1 = 0.93$, $r_2 = 1.2$ and K = 10. If the initial value starting from the inside will eventually tend to the straight line composed of the internal equilibrium family of (3.1), then the boundary equilibria E_1 and E_2 are unstable. However, when the initial value starts from the boundary, the solution starting from the boundary eventually tends to the boundary equilibria. Since the forward invariant domain of (3.1) does not contain boundaries, the boundary equilibria of (3.1) are unstable. Therefore, the boundary equilibria of (A.2) are unstable in $\Phi(\Omega)$.

3.2. Solution curve of the model

In this subsection, we validate the results of the theoretical analysis by plotting the solution curve of (A.1) and marking the corresponding situations for Case 1 to Case 4 in the solution curve. We select the initial value $x_1(0) = 0.005$, $x_2(0) = 0.005$, $Q_1(0) = 0.001$ and $Q_2(0) = 0.001$. In Figures 2–4, the blue dot represents the solution of (A.1) in Case 1, the red dot represents the solution of (A.1) in Case 2, the green dot represents the solution of (A.1) in Case 3 and the black dot represents the solution of (A.1) in Case 4.

Figure 2 shows that, initially, the biomasses of the two plants increased. When $t \in (0, 11)$, the solution of (A.1) corresponds to the solution in Case 4. When t = 11, the solution of (A.1) changes from Case 4 to Case 2. When t = 12, the solution of (A.1) changes from Case 2 to Case 1. Finally, the solution of (A.1) converges to a stable state. This result confirms the conclusion that the internal equilibrium E_* in Case 1 is globally asymptotically stable.



Figure 2. Solution curve of (A.1) when $r_1 = 0.93$, $r_2 = 1.2$, $q_1 = 0.003$, $q_2 = 0.003$, $c_1 = 0.2$, $c_2 = 0.18$, $d_1 = 0.05$, $d_2 = 0.06$, T = 0.03 and K = 8.

Figure 3 shows that the biomasses of two plants increasing over time. When $t \in (0, 11)$, the solution of (A.1) corresponds to the solution in Case 4. When t = 10, the solution of (A.1) transitions from Case 4 to Case 3. When t = 11, the solution of (A.1) transitions from Case 3 to Case 1. When t = 49, the solution of (A.1) transitions from Case 1 to Case 2. Finally, the solution of (A.1) converges to

a stable state. This result verifies the conclusion that the internal equilibrium E_*^1 in Case 2 is locally asymptotically stable.



Figure 3. Solution curve of (A.1) when $r_1 = 1.2$, $r_2 = 0.93$, $q_1 = 0.003$, $q_2 = 0.003$, $c_1 = 0.2$, $c_2 = 0.18$, $d_1 = 0.05$, $d_2 = 0.06$, T = 0.03 and K = 8.

Figure 4 shows that the biomasses of two plants increases over time. When $t \in (0, 10)$, the solution of (A.1) corresponds to the solution in Case 4. When t = 10, the solution of (A.1) changes from Case 4 to Case 2. When t = 11, the solution of (A.1) changes from Case 2 to Case 1. When t = 49, the solution of (A.1) changes from Case 3. Finally, the solution of (A.1) converges to a stable state. This result verifies the conclusion that the internal equilibrium E_*^2 in Case 3 is locally asymptotically stable.

According to Figures 2–4, it can be observed that the solution of (A.1) will appear in three stable states (Cases 1–3). However, no stable solution appears in Case 4, which is consistent with the theoretical analysis above.

3.3. Impact of light intensity and total phosphorus in the system on competition dynamics

In this subsection, we investigate how the light intensity (*K*) and the total phosphorus in the system (*T*) affect the competition outcomes by plotting bifurcation diagrams of (A.1), with respect to *K* and *T*. The phosphorus uptake rates are chosen as $g_i(y) = c_i y$ (i = 1, 2), where $y = T - p_1 - p_2$. Figure 5



Figure 4. Solution curve of (A.1) when $r_1 = 0.93$, $r_2 = 1.2$, $q_1 = 0.003$, $q_2 = 0.003$, $c_1 = 0.18$, $c_2 = 0.2$, $d_1 = 0.06$, $d_2 = 0.05$, T = 0.03 and K = 8.

presents the bifurcation diagrams where the densities of both plants are plotted along the gradient of K. In the bifurcation diagrams, the solid lines represent stable equilibria and the dashed lines signify unstable equilibria. *BP* means a bifurcation point, at which the existence or stability of the equilibria will be transformed. From Figure 5, one observes that when T = 0.03, $K \in (0, 8.61)$, the internal equilibrium E_* is stable and other equilibria (E_0, E_1, E_2) are unstable. This means that these two plants can coexist, even if the light is low ($K \in (0, 8.61)$), when the total phosphorus in the system is adequate. Then, at K = 8.61 (point *BP*), the transcritical bifurcation occurs; that is, E_* changes its stability from being stable to unstable, while the boundary equilibrium E_2 changes from being unstable to stable. When K further increases (K > 8.61), there is only one stable equilibrium E_2 . It indicates that only plant one can survive. The reason for this phenomenon is that the biomasses of both plants increase with the increase of light intensity (K), and there is not enough phosphorus in the ecosystem to support the growth of both plants. Hence, the plant with the lower phosphorus loss rate can adapt to the low total phosphorus in the system conditions and win the competition, and, thus eventually survives.

In particular, through Figure 6, we find that when the parameters of both plants are equal the model has no bifurcation and there is a unique stable equilibrium E_* . This means that two plants will coexist with the same phosphorus loss rate. The biomasses of plants will increase with the increase of light.



Figure 5. Bifurcation diagrams for (A.1) with respect to *K*. (a) the density of plant one *vs*. *K*; (b) the density of plant two *vs*. *K*; (c) the cell quota of plant one *vs*. *K*; (d) the cell quota of plant two *vs*. *K*. The solid line signifies stable equilibria and dashed lines represent unstable equilibria. Here, T = 0.03, $d_1 < d_2$ and the values of other parameters are defined in Table 1. *BP* represents the bifurcation point.

Due to the total phosphorus constraints in the system, the biomasses of both plants will not continue increasing with the increase of light.

From Figure 7, we can observe that when K = 0.25, $T \in (0, 0.735 \times 10^{-3})$, the boundary equilibrium E_0 is stable and other equilibria (E_*, E_1, E_2) are unstable. Hence, both plants will go extinct under low total phosphorus in the system $(T \in (0, 0.735 \times 10^{-3}))$. Then, at $T = 0.735 \times 10^{-3}$ (point *BP*), the transcritical bifurcation occurs, where E_0 changes from being stable to unstable and E_2 changes from being unstable to stable. It indicates that only plant one can survive. This is because the biomasses of both plants increase with the increase of total phosphorus in the system (T), and the plant with the lower phosphorus loss rates has a competitive advantage and eventually survives. When $T \approx 0.98 \times 10^{-3}$ (point *BP*), the transcritical bifurcation occurs $(E_1$ becomes unstable). When T continues increasing until $T \approx 1.86 \times 10^{-3}$ (point *BP*), a transcritical bifurcation occurs, where E_2 changes its stability from being stable to unstable and E_* changes from being unstable to stable. When T further increases $(T > 1.86 \times 10^{-3})$, there is only one stable equilibrium E_* . This indicates that the two plants will coexist



Figure 6. Bifurcation diagrams for (A.1) with respect to *K*. (a) the density of plant one *vs*. *K*; (b) the density of plant two *vs*. *K*; (c) the cell quota of plant one *vs*. *K*; (d) the cell quota of plant two *vs*. *K*. The solid line signifies stable equilibria and dash lines represent unstable equilibria. Here, T = 0.03, $d_1 = d_2 = 0.05$, $r_1 = r_2 = 1.2$, $c_1 = c_2 = 0.2$ and $q_1 = q_2 = 0.003$.

as adequate phosphorus. To conclude, when the light intensity is at a low level, low total phosphorus in the system ($T \in (0, 0.735 \times 10^{-3})$) will lead to the extinction of both plants, and when the total phosphorus in the system is at a moderate level ($0 \le 0.735 \times 10^{-3} \le 0.98 \times 10^{-3}$), only the plant with the lower phosphorus loss rate will survive. When the total phosphorus in the system is sufficient, both plants can survive and coexist ($0.98 \times 10^{-3} \le T \le 1.86 \times 10^{-3}$). However, the biomasses of both plants no longer increases because of the limitation of light intensity.

3.4. Impact of interspecific competition on competition dynamics

In this subsection, we consider the interspecific competition between the two plants and discuss the effect of competition coefficients (b_{12}, b_{21}) on the dynamics of (2.2), and we expound the details for the high, intermediate and low level of total phosphorus in the system with T = 0.03, T = 0.001 and T = 0.0009, respectively. We choose $g_i(y)$ as Michaelis-Menten (Monod) function; that is, $g_i(y) = \frac{c_i y}{a_i + y}$ (i = 1, 2) in (2.2), where $y = T - p_1 - p_2$.

For T = 0.03, we observe from Figure 8(a) that the dynamics are similar to those of Lotka-Volterra



Figure 7. Bifurcation diagrams for (A.1) with respect to *T*. (a) the density of plant one *vs*. *T*; (b) the density of plant two *vs*. *T*; (c) the cell quota of plant one *vs*. *T*; (d) the cell quota of plant two *vs*. *T*. The solid line signifies stable equilibria and dashed lines represent unstable equilibria. Here, K = 0.25, $d_1 < d_2$ and the values of other parameters are defined in Table 1. *BP* represents the bifurcation point.

competition model [1]. That is, when $b_{12} < 1$ and $b_{21} < 1$, the competition between the two plants is not intense and both plants coexist (E_* is stable); when $b_{12} < 1$, $b_{21} > 1$, x_2 goes extinct, x_1 excludes x_2 during competition (E_1 is stable); when $b_{12} > 1$, $b_{21} < 1$, x_1 goes extinct, x_2 excludes x_1 during competition (E_2 is stable); when $b_{12} > 1$, $b_{21} < 1$, x_1 goes extinct, x_2 excludes x_1 equilibria (E_1 and E_2). Which plant wins the competition depends on the initial values of the two plants.

When *T* is decreased to 0.001 (see Figure 8(b)), the dynamics in the region $0 < b_{12} < 1$ and $0 < b_{21} < 1$ are the same as those in Figure 8, implying that the coexistent regions of these two plants are the same. However, the stable regions of E_1 and E_2 are both larger than that in Figure 8(a), and the area of the bistability of E_1 and E_2 is smaller than that with the adequate total phosphorus in the system. Note that b_{12} and b_{21} are the competition coefficients of the two plants, which represent the ability of competing against light, and the higher the value of the two coefficients indicates greater intensity of competition. When the total phosphorus in the system is limited and the competition intensity is high,

the survival of the plants depends more on their own competition ability for light than on the initial state of the plants. For example, the species with greater competitive ability $(b_{12} > b_{21} > 1)$ (plant two) will always win the competition, while the other plant becomes extinct (i.e., E_2 is stable).

As is shown in Figure 8(c), when T is further decreased to 0.0009, the coexistent region of the two plants is larger than that when T = 0.03 and T = 0.001, while there are also two areas of bistability (E_1 and E_* , E_2 and E_*). This suggests that when the total phosphorus in the system is insufficient, less intense competition for light is more favorable for the coexistence of the two plants. Moreover, the limitation of nutrient elements also makes the system unstable, and bistability occurs. In addition, the area of bistability of E_1 and E_2 is smaller than that in Figure 6. This further confirms our findings that when the total phosphorus in the system is limited, the survival of plants is more dependent on their competition ability for light.



Figure 8. Subfigures (a), (b) and (c) show the parameter space corresponding to four conditions of (2.2) with T = 0.003, T = 0.001 and T = 0.0009, respectively. Parameters: $r_1 = r_2 = 1.2$, $q_1 = q_2 = 0.0005$, $c_1 = c_2 = 0.2$, $d_1 = d_2 = 0.05$, $a_1 = a_2 = 0.01$ and K = 2.0.



Figure 9. Bifurcation diagrams for (A.1) with respect to *K* and *T*. (a) the density of plant one *vs*. *K* and *T*; (b) the density of plant two *vs*. *K* and *T*; (c) the cell quota of plant one *vs*. *T* and *K*; (d) the cell quota of plant two *vs*. *K* and *T*. The solid line signifies stable equilibria and dashed lines represent unstable equilibria. Here, $b_{12} < b_{21}$ and the values of other parameters are defined in Table 1.

3.5. Impact of light intensity and total phosphorus in the system on competition dynamics

In this subsection, we investigate how the light intensity (*K*) and the total phosphorus in the system (*T*) affect the competition outcomes by plotting bifurcation diagrams of (A.1) with respect to *K* and *T*. The phosphorus uptake rates are chosen as $g_i(y) = \frac{c_i y}{a_i + y}$ (*i* = 1, 2), where $y = T - p_1 - p_2$. Figure 9 presents the bifurcation diagrams where the densities of both plants are plotted along the gradients of *K* and *T*.

Figure 9 depicts the scenario that the influence coefficient of plant one on plant two is less than the influence coefficient of plant two on plant one, i.e., $b_{12} < b_{21}$. In this condition, plant one will always survive, and its biomass is the highest when both light intensity and total phosphorus are high. Plant two is quite different; the biomass of plant two tends to go extinct even when light intensity is high, and its biomass reaches a maximum when light intensity is low and total phosphorus is high (see Figure 9(a),(b)). In addition, the cell quotas of both plants will tend to be zero with the increase of light



Figure 10. Bifurcation diagrams for (A.1) with respect to *K* and *T*. (a) the density of plant one *vs*. *K* and *T*; (b) the density of plant two *vs*. *K* and *T*; (c) the cell quota of plant one *vs*. *T* and *K*; (d) the cell quota of plant two *vs*. *K* and *T*. The solid line signifies stable equilibria and dashed lines represent unstable equilibria. Here, $b_{12} > b_{21}$ and the values of other parameters are defined in Table 1.

intensity (see Figure 9(c),(d)). This is because plant one has a higher competition ability for light and a lower phosphorus loss rate compared to plant two ($b_{12} < b_{21}$, $d_1 < d_2$). As a result, when light intensity increases and phosphorus becomes limited, plant one gains a competitive advantage, while plant two suffers from a light and nutrient deficiency, leading to its extinction. The cell quotas of both plant one and plant two tends to be zero due to excessive light intensity and insufficient phosphorus. Hence, even though the biomass of plants remains high under phosphorus limitation, the available "nutrition" for their predators is insufficient, which is consistent with the paradox of enrichment.

Similarly, Figure 10 illustrates the changes in plant biomasses and quotas with *K* and *T* when $b_{12} > b_{21}$. From Figure 10 (a),(b), it can be observed that plant one and plant two alternate in their competitive dominance. When light intensity is low, plant two is more likely to dominate, and with increasing *T*, the dominant range of *K* expands. On the other hand, when light intensity is high, plant one is more likely to dominate. This could be because plant two has a higher competitive ability for light compared to plant one. Therefore, when light intensity is low, plant two is capable of absorbing

more light and gaining a competitive advantage. However, as light intensity increases and phosphorus becomes limiting, the lower phosphorus loss rate of plant one helps it retain more nutrients and allows it to dominate in the competition.

In summary, the light intensity (*K*), total phosphorus (*T*), interspecific competition ability (b_{12}, b_{21}) and loss rate of phosphorus (d_1, d_2) all affect the coexistence and competition of two plants. Additionally, excessive light intensity may reduce the cell quotas for both plants, which is in good agreement with the findings of the stoichiometric competition model involving two predators and one prey [26], where both predators were limited by the prey's phosphorus content.

4. Discussion

In this study, we established a stoichiometric competition model between two plants that explicitly incorporated the impact of the element cycle on plants. We systematically discussed the effects of light intensity and total phosphorus in the system on the dynamics of the model.

In mathematical analysis, when the competition for lights between two plants is equal with $b_{12} = b_{21}$, we give the existence and stability analysis of the equilibria of this model in four conditions. When the growth of two plants is only limited by light intensity, the two boundary equilibria are unstable and the coexistence equilibrium is globally asymptotically stable. When the growth of plant one is limited by light intensity and plant two is limited by the total phosphorus in the system, the boundary equilibrium is unstable and the coexistence equilibrium is locally asymptotically stable if $d_2 < \frac{c_2 d_1}{c_1}$. When the growth of plant one is limited by the total phosphorus in the system and plant two is limited by light intensity, the boundary equilibrium is unstable and the coexistence equilibrium is unstable and the coexistence equilibrium is locally asymptotically stable if $d_1 < \frac{c_1 d_2}{c_2}$. When the growth of two plants are only limited by the total phosphorus in the system, the model does not have coexistence equilibrium and we give the conditions for the existence and stability of boundary equilibria.

When the competition for lights between two plants is equal with $b_{12} = b_{21}$, the numerical results indicate that light intensity and total phosphorus in the system play important roles in the growth and coexistence of two plants. When the total phosphorus in the system T is fixed, moderate light intensity and sufficient total phosphorus in the system are beneficial for the coexistence of two plants. However, excessive light intensity can disrupt this kind of balance. Due to the limitation of the total phosphorus in the system, the loss rate of phosphorus determines the outcome of competition. Plants with a lower phosphorus loss rate will win the competition. This means that a higher light intensity can help plants with lower phosphorus loss rates win. In particular, when the parameters of both plants are equal, the two plants will coexist under sufficient light and nutrients. Moreover, as the light intensity increases, the quality of both plants will deteriorate. When the light intensity K is fixed, two plants will go extinct due to the low total phosphorus in the system. With the enhancement of the total phosphorus in the system, the plant with the lower phosphorus loss rate can survive. This means that both plants cannot survive with lower total phosphorus in the system, while moderate total phosphorus in the system can help plants with lower phosphorus loss rates win. As the total phosphorus in the system further increases, the two plants will coexist. However, due to light limitation, the biomasses of the two plants will not continue increasing. At the same time, the increase in nutrients will improve the quality of two plants.

When the competition for light between two plants is not equal with $b_{12} \neq b_{21}$, the numerical

simulation results also show that for a high level of total phosphorus in the system (T = 0.03), the dynamic properties of our model are the same as those in [1]. That is, when two plants have different competitive abilities the model exhibits bistability (E_1 and E_2 are stable). At this time, the competition results of the two plants depend on their initial values. For a moderate level of the total phosphorus in the system (T = 0.001), the bistable region of E_1 and E_2 will decrease. When two plants have a great competition ability, the competitive outcome of the two plants depends on their ability to compete for light. At a low level of total phosphorus in the system (T = 0.0009), the model becomes unstable, resulting in more types of bistability (E_1 and E_* are stable, E_2 and E_* are stable). At this time, the bistable region of E_1 and E_2 continues decreasing, and the coexistence area of the two plants becomes larger. Meanwhile, the survival of the two plants depends on their competition for light; the smaller the competition for light, the more favorable for the coexistence of the two plants. We conclude that when the total phosphorus in the system is at moderate or low level, competitive results depend on a plant's ability of competing for light rather than its initial state.

Our study still has several limitations. First, in order to facilitate the mathematical analysis of the model, we assumed that $b_{12} = b_{21}$. However, from a biological perspective, equal competition for light (b_{12} equals b_{21}) is improbable, as no two plants possess identical parameters. Second, due to the complexity of the model, we didn't obtain the global asymptotic stability of the coexistence equilibrium. Third, for the mathematical analysis, we assumed that the phosphorus uptake rate function was linear, while for the nonlinear scenario, we used numerical simulations to investigate the dynamics of the model, which will be studied in future work.

Use of AI tools declaration

The authors declare they have not used Artificial Intelligence (AI) tools in the creation of this article.

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Conflict of interest

The authors declare that there is no conflict of interest.

References

- 1. J. D. Murray, Mathematical biology: I. An introduction, Springer, 17 (2002).
- Y. Song, M. Han, Y. Peng, Stability and hopf bifurcations in a competitive lotkavolterra system with two delays, *Chaos Solitons Fractals*, 22 (2004), 1139–1148. https://doi.org/10.1016/j.chaos.2004.03.026

- 3. Z. Jin, H. L. Guihua, The persistence in a lotka-volterra competition Maoan, with impulsive, Solitons (2005),systems Chaos Fractals. 24 1105-1117. https://doi.org/10.1016/j.chaos.2004.09.065
- X. Zhang, L. Chen, The linear and nonlinear diffusion of the competitive lotkavolterra model, Nonlinear Analysis: Theory, *Methods Appl.*, 66 (2007), 2767–2776. https://doi.org/10.1016/j.na.2006.04.006
- 5. S. B. Hsu, X. Q. Zhao, A lotka-volterra competition model with seasonal succession, *J. Math. Biol.*, **64** (2012), 109–130. https://doi.org/10.1007/s00285-011-0408-6
- 6. M. Wang, J. Zhao, Free boundary problems for a lotka-volterra competition system, *J. Dyn. Differ. Equations*, **26** (2014) 655-672. https://doi.org/10.1007/s10884-014-9363-4
- 7. M. Liu, M. Fan, Permanence of stochastic lotka-volterra systems, J. Nonlinear Sci., 27 (2017), 425–452. https://doi.org/10.1007/s00332-016-9337-2
- 8. G. Ren, B. Liu, Global solvability and asymptotic behavior in a two-species chemotaxis system with lotka-volterra competitive kinetics, *Math. Models Methods Appl. Sci.*, **31** (2021), 941–978. https://doi.org/10.1142/S0218202521500238
- 9. I. Loladze, Y. Kuang, J. J. Elser, Stoichiometry in producer-grazer systems: linking energy flow with element cycling, *Bull. Math. Biol.*, **62** (2000), 1137–1162. https://doi.org/10.1006/bulm.2000.0201
- 10. R. W. Sterner, J. J. Elser, *Ecological stoichiometry, in: Ecological Stoichiometry*, Princeton University Press, 2017. https://doi.org/10.1515/9781400885695
- 11. X. Yang, X. Li, H. Wang, Y. Kuang, Stability and bifurcation in a stoichiometric producer-grazer model with knife edge, *SIAM J. Appl. Dyn. Syst.*, **15** (2016), 2051–2077. https://doi.org/10.1137/15M1023610
- 12. C. M. Davies, H. Wang, Incorporating carbon dioxide into a stoichiometric producer-grazer model, *J. Math. Biol.*, **83** (2021), 1–48. https://doi.org/10.1007/s00285-021-01658-3
- J. Zhang, J. D. Kong, J. Shi, H. Wang, Phytoplankton competition for nutrients and light in a stratified lake: a mathematical model connecting epilimnion and hypolimnion, *J. Nonlinear Sci.*, **31** (2021), 1–42. https://doi.org/10.1007/s00332-021-09693-6
- 14. H. Wang, Z. Lu, A. Raghavan, Weak dynamical threshold for the "strict homeostasis" assumption in ecological stoichiometry, *Ecol. Model.*, **384** (2018), 233–240. https://doi.org/10.1016/j.ecolmodel.2018.06.027
- 15. X. Li, H. Wang, A stoichiometrically derived algal growth model and its global analysis, *Math. Biosci. Eng.*, **7** (2010), 825–836. https://doi.org/10.3934/mbe.2010.7.825
- J. D. Kong, H. Wang, T. Siddique, J. Foght, K. Semple, Z. Burkus, et al., Second-generation stoichiometric mathematical model to predict methane emissions from oil sands tailings, *Sci. Total Environ.*, 694 (2019), 133645. https://doi.org/10.1016/j.scitotenv.2019.133645

- V. Kirkow, H. Wang, P. V. Garcia, S. Ahmed, C. M. Heggerud, Impacts of a changing environment on a stoichiometric producer-grazer system: a stochastic modelling approach, *Ecol. Model.*, 469 (2022), 109971. https://doi.org/10.1016/j.ecolmodel.2022.109971
- H. Wang, Y. Kuang, I. Loladze, Dynamics of a mechanistically derived stoichiometric producergrazer model, J. Biol. Dyn., 2 (2008), 286–296. https://doi.org/10.1080/17513750701769881
- A. Peace, H. Wang, Y. Kuang, Dynamics of a producer-grazer model incorporating the effects of excess food nutrient content on grazer's growth, *Bull. Math. Biol.*, **76** (2014), 2175–2197. https://doi.org/10.1007/s11538-014-0006-z
- 20. D. Tilman, Mechanisms of plant competition for nutrients: the elements of a predictive theory of competition, *Perspectives on Plant Competition*, Academic Press, (1990), 117–141.
- 21. J. Ji, H. Wang, Competitive exclusion and coexistence in a stoichiometric chemostat model, *J. Dyn. Differ. Equations*, 2022. https://doi.org/10.1007/s10884-022-10188-5
- Y. Kuang, J. Huisman, J. J. Elser, Stoichiometric plant-herbivore models and their interpretation, *Math. Biosci. Eng.*, 1 (2004) 215-222. https://doi.org/10.3934/mbe.2004.1.215
- L. Markus, II. asymptotically autonomous differential systems, Contributions to the Theory of Nonlinear Oscillations, Princeton University Press, 3 (2016), 17. https://doi.org/10.1515/9781400882175-003
- 24. A. Hurwitz. Ueber die Bedingungen, unter welchen eine Gleichung nur Wurzeln mit negativen reellen Theilen besitzt, *Mathematische Annalen*, **46** (1895), 273–284.
- 25. T. A. Burton, Volterra integral and differential equations, Elsevier, 2005.
- I. Loladze, Y. Kuang, J. J. Elser, W. F. Fagan, Competition and stoichiometry: coexistence of two predators on one prey, *Theor. Popul. Biol.*, 65 (2004), 1–15. https://doi.org/10.1016/S0040-5809(03)00105-9

Appendix A. Mathematical analysis of (2.4)

In this section, we verify the dissipativity of (2.4), then derive the equilibria and analyze their stabilities. For the convenience of subsequent analysis of the model, we let $b_{12} = b_{21} = 1$.

Theorem 1. The solutions with initial conditions in the set

$$\Phi(\Omega) = \left\{ (x_1, x_2, Q_1, Q_2) : 0 < x_1 < \min\{K, T/q_1\}, 0 < x_2 < \min\{K, T/q_2\}, Q_1 > q_1, Q_2 > q_2, Q_1 > q_1, Q_2 > q_1, Q_1 > q_1, Q_2 > q_1, Q_1 >$$

remain there for all forward times.

Proof We consider a solution $X(t) = (x_1(t), x_2(t), Q_1(t), Q_2(t))$ of (2.2) with initial condition in $\Phi(\Omega)$. Hence, $0 < x_1(0) < \min\{K, T/q_1\}, 0 < x_2(0) < \min\{K, T/q_2\}, Q_1(0) > q_1, Q_2(0) > q_2, Q_1(0)x_1(0) + Q_2(0)x_2(0) < T$. We assume that $\exists t_1 > 0$, such that X(t) touches or crosses the boundary of $\overline{\Omega}$ (closure of $\Phi(\Omega)$ for the first time, then $(x_1(t), x_2(t), Q_1(t), Q_2(t)) \in \Phi(\Omega)$ for $0 \le t < t_1$. We discuss the following cases:

(*i*) $x_1(t_1) = 0$. $Q_1(t) > q_1$, $Q_2(t) > q_2$, $Q_1(t)x_1(t) + Q_2(t)x_2(t) < T$ for $0 \le t \le t_1$. Let $a_1 = \min\{Q_1(t) : t \in [0, t_1]\} > 0$. Then, for $0 \le t \le t_1$, we have

$$\begin{aligned} \frac{dx_1}{dt} &= r_1 x_1 \min\left\{1 - \frac{x_1 + x_2}{K}, 1 - \frac{q_1}{Q_1}\right\} \\ &\geq r_1 x_1 \min\left\{1 - \frac{\min\{K, T/q_1\} + \min\{K, T/q_2\}}{K}, 1 - \frac{q_1}{a_1}\right\} \\ &= \left[r_1 \min\left\{1 - \frac{\min\{K, T/q_1\} + \min\{K, T/q_2\}}{K}, 1 - \frac{q_1}{a_1}\right\}\right] x_1 \equiv \mu x_1, \end{aligned}$$

(*ii*) $x_2(t_1) = 0$. The proof is the same as that for (*i*).

(*iii*) $x_1(t_1) = \min\{K, T/q_1\}$. $0 < x_2(t) < \min\{k, T/q_2\}$, $Q_1(t)x_1(t) + Q_2(t)x_2(t) \le T$ for $0 \le t \le t_1$ implies $Q_1(t)x_1(t) \le T$ for $0 \le t \le t_1$. Then, for $0 \le t \le t_1$, we have

$$\frac{dx_1}{dt} = r_1 x_1 \min\left\{1 - \frac{x_1 + x_2}{K}, 1 - \frac{q_1}{Q_1}\right\},\,$$

if $\frac{dx_1}{dt} = r_1(1 - \frac{x_1 + x_2}{K})$, then $x_1(t) + x_2(t) < K$, $x_1(t) < K$. Moreover, $Q_1(t)x_1(t) \le T$, then $x_1(t) < T/q_1$. Therefore, $x_1 < \min\{K, T/q_1\}$ for all $0 \le t \le t_1$, which is a contradiction. (*iv*) $x_2(t_1) = \min\{K, T/q_2\}$. The proof is the same as that for (*iii*).

(v) $Q_1(t_1) = q_1$. $Q(t) = Q_1(t) - q_1$ then $Q(t_1) = 0$ and Q(t) > 0 for $0 \le t < t_1$. For $0 \le t \le t_1$, we have

$$\frac{dQ}{dt} = g_1(T - Q_1x_1 - Q_2x_2) - d_1Q_1 - Q_1r_1\min\left\{1 - \frac{x_1 + x_2}{K}, 1 - \frac{q_1}{Q_1}\right\},\\ \ge -d_1Q_1 - Q_1a,$$

where *a* is a constant. Thus, $Q(t) \ge Q_1(0)e^{-d_1-at} > 0$ for $0 \le t \le t_1$, which is a contradiction. (*vi*) $Q_2(t_1) = q_2$. The proof is the same as that for (*v*).

(vii) $Q_1(t_1)x_1(t_1) + Q_2(t_1)x_2(t_1) = T$. $z(t) = T - Q_1(t)x_1(t) + Q_2(t)x_2(t)$ then $z(t_1) = 0$ and z(t) > 0 for $0 \le t < t_1$. Then, for $0 \le t \le t_1$, we have

$$\begin{aligned} \frac{dz}{dt} &= -\frac{dQ_1}{dt} x_1 - \frac{dx_1}{dt} Q_1 - \frac{dQ_2}{dt} x_2 - \frac{dx_2}{dt} Q_2 \\ &= -g_1(z) x_1 - g_2(z) x_2 + d_1 Q_1 x_1 + d_2 Q_2 x_2 \\ &\ge -g_1'(0) z \min\{K, T/q_1\} - g_2'(0) z \min\{K, T/q_2\} \\ &= -[g_1'(0) \min\{K, T/q_1\} - g_2'(0) \min\{K, T/q_2\}] z \equiv -\tilde{v}z, \end{aligned}$$

where $\tilde{v} > 0$ is a constant. Thus, $z(t) \ge z(0)e^{-\tilde{v}t} > 0$ for $0 \le t \le t_1$, which is a contradiction. \Box

For simplicity, we assume that $g_i(x) = c_i x$, i = 1, 2. In this case, (2.4) becomes

$$\begin{cases} \frac{dx_1}{dt} = r_1 x_1 \min\left\{1 - \frac{x_1 + x_2}{K}, 1 - \frac{q_1}{Q_1}\right\}, \\ \frac{dx_2}{dt} = r_2 x_2 \min\left\{1 - \frac{x_1 + x_2}{K}, 1 - \frac{q_2}{Q_2}\right\}, \\ \frac{dQ_1}{dt} = c_1 (T - Q_1 x_1 - Q_2 x_2) - d_1 Q_1 - Q_1 r_1 \min\left\{1 - \frac{x_1 + x_2}{K}, 1 - \frac{q_1}{Q_1}\right\}, \\ \frac{dQ_2}{dt} = c_2 (T - Q_1 x_1 - Q_2 x_2) - d_2 Q_2 - Q_2 r_2 \min\left\{1 - \frac{x_1 + x_2}{K}, 1 - \frac{q_2}{Q_2}\right\}. \end{cases}$$
(A.1)

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We analyze the stability of the equilibrium of (A.1) in the following four cases.

Case 1.
$$\frac{x_1 + x_2}{K} > \frac{q_1}{Q_1}, \frac{x_1 + x_2}{K} > \frac{q_2}{Q_2}$$

$$\begin{cases} \frac{dx_1}{dt} = r_1 x_1 \left(1 - \frac{x_1 + x_2}{K}\right), \\ \frac{dx_2}{dt} = r_2 x_2 \left(1 - \frac{x_1 + x_2}{K}\right), \\ \frac{dQ_1}{dt} = c_1 (T - Q_1 x_1 - Q_2 x_2) - d_1 Q_1 - Q_1 r_1 \left(1 - \frac{x_1 + x_2}{K}\right), \\ \frac{dQ_2}{dt} = c_2 (T - Q_1 x_1 - Q_2 x_2) - d_2 Q_2 - Q_2 r_2 \left(1 - \frac{x_1 + x_2}{K}\right). \end{cases}$$
(A.2)

In Case 1, the boundary equilibria are $E_1 = (0, K, \frac{c_1 T d_2}{d_1(c_2 K + d_2)}, \frac{c_2 T}{c_2 K + d_2})$ and $E_2 = (K, 0, \frac{c_1 T}{c_1 K + d_1}, \frac{c_2 T d_1}{d_2(c_1 K + d_1)})$. The coexistence equilibrium is $E_* = (\tilde{x}_1, \tilde{x}_2, \tilde{Q}_1, \tilde{Q}_2)$, where

$$\tilde{x}_1 = \frac{d_1 \tilde{Q}_1 + c_1 \tilde{Q}_2 K - c_1 T}{c_1 (\tilde{Q}_2 - \tilde{Q}_1)}, \quad \tilde{x}_2 = \frac{d_2 \tilde{Q}_2 + c_2 \tilde{Q}_1 K - c_2 T}{c_2 (\tilde{Q}_1 - \tilde{Q}_2)},$$
$$\tilde{x}_1 + \tilde{x}_2 = K.$$

The Jacobian matrix at (x_1, x_2, Q_1, Q_2) is

$$J(x_1, x_2, Q_1, Q_2) = \begin{pmatrix} r_1 \left(1 - \frac{2x_1 + x_2}{K}\right) & -\frac{r_1 x_1}{K} & 0 & 0\\ -\frac{r_2 x_2}{K} & r_2 \left(1 - \frac{2x_2 + x_1}{K}\right) & 0 & 0\\ Q_1 \left(\frac{r_1}{K} - c_1\right) & \frac{Q_1 r_1}{K} - c_1 Q_2 & M & -c_1 x_2\\ \frac{Q_2 r_2}{K} - c_2 Q_1 & Q_2 \left(\frac{r_2}{K} - c_2\right) & -c_2 x_1 & N \end{pmatrix}$$

where

$$M = \frac{r_1(x_1 + x_2)}{K} - c_1 x_1 - d_1 - r_1,$$

$$N = \frac{r_2(x_1 + x_2)}{K} - c_2 x_2 - d_2 - r_2.$$

The boundary equilibria, E_1 and E_2 , are unstable and their stability is complex. We will discuss their stability in the numerical simulation.

Next, we explore the globally asymptotic stability of E_* . The main approach involves the theory of asymptotic autonomous systems and Dulac's criterion [25]. We shall first summarize a few concepts and results of asymptotic autonomous system from Markus (2016) [12]. Consider the following differential equations

$$\frac{dx}{dt} = f(t, x), \tag{A.3}$$

$$\frac{dy}{dt} = g(y). \tag{A.4}$$

Equation (A.3) is asymptotically autonomous with limit equation (A.4) if

$$f(t, x) \to g(x), t \to \infty$$
, locally uniformly in $x \in \mathbb{R}^n$,

i.e., for x in any compact subset of R^n . Assume that f(t, x), g(x) are continuous functions and locally Lipschitz in x.

Lemma 1. ([23]) The ω -limit set Ω of a forward bounded solution x to (A.3) is not empty, compact and connected. Moreover, Ω attracts x, i.e.,

$$dist(x(t), \Omega) \to 0, t \to \infty.$$

Hence, Ω is invariant under (A.3). In particular, any point in Ω lies on an orbit of (A.4) in Ω .

Lemma 2. ([23]) Let \hat{E} be a locally asymptotically stable equilibrium of (A.4) and let Ω be the ω -limit set of forward bounded solution x of (A.3). If Ω contains a point y_0 , such that the solution y of (A.4) with $y(0) = y_0$ converges to \hat{E} when $t \to \infty$. Then, $\Omega = \{\hat{E}\}$, i.e., $x(t) \to \hat{E}$ when $t \to \infty$.

The existence and stability of coexistence equilibrium $E_* = (\tilde{x}_1, \tilde{x}_2, \tilde{Q}_1, \tilde{Q}_2)$ is analyzed below, where

$$\tilde{x}_1 = \frac{d_1 \tilde{Q}_1 + c_1 \tilde{Q}_2 K - c_1 T}{c_1 (\tilde{Q}_2 - \tilde{Q}_1)}, \quad \tilde{x}_2 = \frac{d_2 \tilde{Q}_2 + c_2 \tilde{Q}_1 K - c_2 T}{c_2 (\tilde{Q}_1 - \tilde{Q}_2)},$$
$$\tilde{x}_1 + \tilde{x}_2 = K.$$

Theorem 2. If $\tilde{Q_1} > q_1$, $\tilde{Q_2} > q_2$ and $\frac{c_2 d_1 \tilde{Q_1}^2 - c_1 d_2 \tilde{Q_2}^2}{c_1 c_2 (\tilde{Q_2} - \tilde{Q_1})} < 0$, E_* exists and it is globally asymptotically stable.

Proof The Jacobian matrix at E_* is

$$J(E_*) = \begin{pmatrix} -\frac{r_1 \tilde{x}_1}{K} & -\frac{r_1 \tilde{x}_1}{K} & 0 & 0\\ -\frac{r_2 \tilde{x}_2}{K} & -\frac{r_2 \tilde{x}_2}{K} & 0 & 0\\ \tilde{Q}_1 \left(\frac{r_1}{K} - c_1\right) & \frac{\tilde{Q}_1 r_1}{K} - c_1 \tilde{Q}_2 & -c_1 \tilde{x}_1 - d_1 & -c_1 \tilde{x}_2\\ \frac{\tilde{Q}_2 r_2}{K} - c_2 \tilde{Q}_1 & \tilde{Q}_2 \left(\frac{r_2}{K} - c_2\right) & -c_2 \tilde{x}_1 & -c_2 \tilde{x}_2 - d_2 \end{pmatrix}.$$

The eigenvalues of $J(E_*)$ are

$$\begin{split} \lambda_1 &= 0, \quad \lambda_2 = -\frac{r_1 \tilde{x}_1 + r_2 \tilde{x}_2}{K} < 0, \\ \lambda_3 &= -\frac{c_1 \tilde{x}_1 + c_2 \tilde{x}_2 + d_1 + d_2}{2} + \frac{\sqrt{(c_1 \tilde{x}_1 + c_2 \tilde{x}_2 + d_1 + d_2)^2 - 4(c_2 \tilde{x}_2 d_1 + c_1 \tilde{x}_1 d_2 + d_1 d_2)}}{2} < 0, \\ \lambda_4 &= -\frac{c_1 \tilde{x}_1 + c_2 \tilde{x}_2 + d_1 + d_2}{2} - \frac{\sqrt{(c_1 \tilde{x}_1 + c_2 \tilde{x}_2 + d_1 + d_2)^2 - 4(c_2 \tilde{x}_2 d_1 + c_1 \tilde{x}_1 d_2 + d_1 d_2)}}{2} < 0. \end{split}$$

We divide (A.2) into two subsystems. Next, we consider the following submodel of (A.2)

$$\begin{cases} \frac{dx_1}{dt} = r_1 x_1 \left(1 - \frac{x_1 + x_2}{K} \right) = f(x_1, x_2), \\ \frac{dx_2}{dt} = r_2 x_2 \left(1 - \frac{x_1 + x_2}{K} \right) = g(x_1, x_2). \end{cases}$$
(A.5)

Let

$$B(x_1, x_2) = \frac{1}{x_1 x_2},$$

then

$$\frac{\partial(Bf)}{\partial x_1} + \frac{\partial(Bg)}{\partial x_2} = -\frac{r_1}{Kx_2} - \frac{r_2}{Kx_1} < 0.$$

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Therefore, (A.5) has no limit cycle by Dulac's criterion [25]. Both boundary equilibria of (A.5) are unstable. Therefore, the solution of (A.5) will tend to the globally asymptotically stable internal equilibrium, implying that the solution of (A.5) will approach the line $x_1 = K - x_2$.

Note that the first two equations of (A.2) are independent on Q_1 and Q_2 . Moreover, $(\tilde{x}_1, \tilde{x}_2)$ is globally asymptotically stable. Next, we analyze the last equations of (A.2)

$$\begin{cases} \frac{dQ_1}{dt} = g_1(T - Q_1 x_1(t) - Q_2 x_2(t)) - d_1 Q_1, \\ \frac{dQ_2}{dt} = g_2(T - Q_1 x_1(t) - Q_2 x_2(t)) - d_2 Q_2. \end{cases}$$
(A.6)

According to the above analysis in (A.5), the solution of the (A.5) will approach the line $x_1 = K - x_2$. Let $\lim_{t\to\infty} x_1(t) = m$, $\lim_{t\to\infty} x_2(t) = K - m = n$. Here, $m < \min\{K, T/q_1\}$ and $n < \min\{K, T/q_2\}$. Moreover, $Q_1m + Q_2n < T$. Then, the limiting system of (A.6) is

$$\begin{cases}
\frac{dQ_1}{dt} = c_1 T - (c_1 m + d_1)Q_1 - c_1 nQ_2 = P_1(Q_1, Q_2), \\
\frac{dQ_2}{dt} = c_2 T - c_2 mQ_1 - (c_2 n + d_2)Q_2 = P_2(Q_1, Q_2).
\end{cases}$$
(A.7)

Let

$$D(x_1, x_2) = \frac{1}{Q_1 Q_2}$$

then

$$\frac{\partial(DP_1)}{\partial Q_1} + \frac{\partial(DP_2)}{\partial Q_2} = \frac{c_1 Q_2 (nQ_2 - T) + c_2 T (mQ_1 - T)}{Q_1^2 Q_2^2}.$$

Clearly, $nQ_2 < T$ and $mQ_1 < T$. It implies that $\frac{\partial(DP_1)}{\partial Q_1} + \frac{\partial(DP_2)}{\partial Q_2} < 0$. Therefore, it can be concluded that (A.7) has no limit cycle by applying Dulac's criterion [25] and the solution of (A.7) will tend to be globally asymptotically stable. This means that the solution of (A.6) will approach the line $Q_1 = \frac{c_1 d_2}{c_2 d_1} Q_2$. By the theory of asymptotic autonomous systems, the internal equilibrium $(\tilde{Q}_1, \tilde{Q}_2)$ of (A.6) is globally asymptotically stable. Hence, (A.2) has no nontrivial periodic solutions in $\Phi(\Omega)$. The internal equilibrium E_* of (A.2) is globally asymptotically stable.

Case 2.
$$\frac{x_1 + x_2}{K} > \frac{q_1}{Q_1}, \frac{x_1 + x_2}{K} < \frac{q_2}{Q_2}$$

$$\begin{cases} \frac{dx_1}{dt} = r_1 x_1 \left(1 - \frac{x_1 + x_2}{K}\right), \\ \frac{dx_2}{dt} = r_2 x_2 \left(1 - \frac{q_2}{Q_2}\right), \\ \frac{dQ_1}{dt} = c_1 (T - Q_1 x_1 - Q_2 x_2) - d_1 Q_1 - Q_1 r_1 \left(1 - \frac{x_1 + x_2}{K}\right), \\ \frac{dQ_2}{dt} = c_2 (T - Q_1 x_1 - Q_2 x_2) - d_2 Q_2 - Q_2 r_2 \left(1 - \frac{q_2}{Q_2}\right). \end{cases}$$
(A.8)

In Case 2, the boundary equilibrium is $E_1^1 = \left(0, \frac{T}{q_2} - \frac{d_2}{c_2}, \frac{c_1 K d_2 q_2^2}{c_2 q_2 K (d_1 + r_1) - r_1 (c_2 T - d_2 q_2)}, q_2\right)$ and the coexistence equilibrium is $E_*^1 = (x_1^{*1}, x_2^{*1}, Q_1^{*1}, Q_2^{*1})$, where

$$x_1^{*1} = K - \frac{d_2 q_2 (d_1 + c_1 K) - c_2 d_1 T}{q_2 (d_2 c_1 - d_1 c_2)}, \quad x_2^{*1} = \frac{d_2 q_2 (d_1 + c_1 K) - c_2 d_1 T}{q_2 (d_2 c_1 - d_1 c_2)},$$
$$Q_1^{*1} = \frac{d_2 q_2 c_1}{c_2 d_1}, \quad Q_2^{*1} = q_2.$$

The Jacobian matrix at (x_1, x_2, Q_1, Q_2) is

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$$J(x_1, x_2, Q_1, Q_2) = \begin{pmatrix} r_1 \left(1 - \frac{2x_1 + x_2}{K}\right) & -\frac{r_1 x_1}{K} & 0 & 0 \\ 0 & r_2 \left(1 - \frac{q_2}{Q_2}\right) & 0 & \frac{r_2 x_2 q_2}{Q_2^2} \\ Q_1 \left(\frac{r_1}{K} - c_1\right) & \frac{Q_1 r_1}{K} - c_1 Q_2 & \frac{r_1 (x_1 + x_2)}{K} - c_1 x_1 - d_1 - r_1 & -c_1 x_2 \\ -c_2 Q_1 & -c_2 Q_2 & -c_2 x_1 & -c_2 x_2 - d_2 - r_2 \end{pmatrix}.$$

Theorem 3. If $0 < \frac{T}{q_2} - \frac{d_2}{c_2} < K$ and $K < \frac{(c_2T - d_2q_2)(c_1d_1q_2 + r_1q_1c_2)}{q_1q_2c_2^2(d_1+r_1)}$, E_1^1 exists and it is unstable.

Proof The Jacobian matrix at E_1^1 is

$$J(E_1^1) = \begin{pmatrix} r_1 \left(1 - \frac{x_2}{K}\right) & 0 & 0 & 0 \\ 0 & 0 & 0 & \frac{r_2 x_2}{q_2} \\ Q_1 \left(\frac{r_1}{K} - c_1\right) & \frac{Q_1 r_1}{K} - c_1 q_2 & \frac{r_1 x_2}{K} - d_1 - r_1 & -c_1 x_2 \\ -c_2 Q_1 & -c_2 q_2 & 0 & -c_2 x_2 - d_2 - r_2 \end{pmatrix}$$

The eigenvalues of $J(E_1^1)$ are

$$\begin{split} \lambda_1 &= r_1 \left(1 - \frac{\frac{T}{q_2} - \frac{d_2}{c_2}}{K} \right) > 0, \lambda_2 = \frac{r_1}{K} \left(\frac{T}{q_2} - \frac{d_2}{c_2} \right) - d_1 - r_1 \\ \lambda_3 &= \frac{-\left(\frac{c_2T}{q_2} + r_2\right) - \sqrt{\left(\frac{c_2T}{q_2} - r_2\right)^2 + 4r_2 d_2}}{2} < 0, \\ \lambda_4 &= \frac{-\left(\frac{c_2T}{q_2} + r_2\right) + \sqrt{\left(\frac{c_2T}{q_2} - r_2\right)^2 + 4r_2 d_2}}{2}. \end{split}$$

Since $\lambda_1 > 0$, E_1^1 is unstable.

The existence and local stability of coexistence equilibrium $E_*^1 = (x_1^{*1}, x_2^{*1}, Q_1^{*1}, Q_2^{*1})$ is analyzed below, where

$$x_1^{*1} = K - \frac{d_2 q_2 (d_1 + c_1 K) - c_2 d_1 T}{q_2 (d_2 c_1 - d_1 c_2)}, \quad x_2^{*1} = \frac{d_2 q_2 (d_1 + c_1 K) - c_2 d_1 T}{q_2 (d_2 c_1 - d_1 c_2)},$$
$$Q_1^{*1} = \frac{d_2 q_2 c_1}{c_2 d_1}, \quad Q_2^{*1} = q_2.$$

Theorem 4. For the case of $d_2 > \frac{c_2 d_1 q_1}{c_1 q_2}$, $0 < \frac{d_2 q_2 (d_1 + c_1 K) - c_2 d_1 T}{q_2 (d_2 c_1 - d_1 c_2)} < K$, $d_2 c_1 - d_1 c_2 \neq 0$, E_*^1 exists. E_*^1 is locally asymptotically stable if $d_2 < \frac{c_2 d_1}{c_1}$.

Proof The Jacobian matrix at E_*^1 is

$$J(E_*^1) = \begin{pmatrix} -\frac{r_1 x_1^{*1}}{K} & -\frac{r_1 x_1^{*1}}{K} & 0 & 0 \\ 0 & 0 & 0 & \frac{r_2 x_2^{*1}}{q_2} \\ Q_1^{*1} \left(\frac{r_1}{K} - c_1\right) & \frac{Q_1^{*1} r_1}{K} - c_1 q_2 & -c_1 x_1^{*1} - d_1 & -c_1 x_2^{*1} \\ -c_2 Q_1^{*1} & -c_2 q_2 & -c_2 x_1^{*1} & -c_2 x_2^{*1} - d_2 - r_2 \end{pmatrix}$$

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The corresponding characteristic equation is

$$f(\lambda) = \lambda^4 + A_1 \lambda^3 + A_2 \lambda^2 + A_3 \lambda + A_4,$$

where

$$\begin{split} A_{1} &= \frac{r_{1}x_{1}^{*1}}{K} + c_{1}x_{1}^{*1} + c_{2}x_{2}^{*1} + d_{1} + d_{2} + r_{2}, \\ A_{2} &= \frac{r_{1}x_{1}^{*1}(c_{1}x_{1}^{*1} + d_{1} + c_{2}x_{2}^{*1} + d_{2} + r_{2})}{K} + c_{1}x_{1}^{*1}d_{2} + c_{1}x_{1}^{*1}r_{2} + c_{2}x_{2}^{*1}d_{1} + c_{2}r_{2}x_{2}^{*1} + d_{1}d_{2} + d_{1}r_{2}, \\ A_{3} &= \frac{r_{1}x_{1}^{*1}(c_{1}x_{1}^{*1}d_{2} + c_{1}x_{1}^{*1}r_{2} + c_{2}x_{2}^{*1}d_{1} + c_{2}r_{2}x_{2}^{*1} + d_{1}d_{2} + d_{1}r_{2})}{K} + r_{2}c_{2}d_{1}x_{2}^{*1}, \\ A_{4} &= \frac{r_{1}r_{2}c_{2}x_{1}^{*1}x_{2}^{*1}d_{1}(q_{2} - Q_{1}^{*1})}{Kq_{2}}. \end{split}$$

We verify that

$$A_1 > 0$$
, $A_1A_2 - A_3 > 0$, and $A_1A_2A_3 - A_1^2A_4 - A_3^2 > 0$.

If $d_2 < \frac{c_2 d_1}{c_1}$, $A_4 > 0$. According to the Routh-Hurwitz criterion [24], E_*^1 is locally asymptotically stable. **Case 3.** $\frac{x_1 + x_2}{K} < \frac{q_1}{Q_1}, \frac{x_1 + x_2}{K} > \frac{q_2}{Q_2}$

$$\begin{cases} \frac{dx_1}{dt} = r_1 x_1 \left(1 - \frac{q_1}{Q_1}\right), \\ \frac{dx_2}{dt} = r_2 x_2 \left(1 - \frac{x_1 + x_2}{K}\right), \\ \frac{dQ_1}{dt} = c_1 (T - Q_1 x_1 - Q_2 x_2) - d_1 Q_1 - Q_1 r_1 \left(1 - \frac{q_1}{Q_1}\right), \\ \frac{dQ_2}{dt} = c_2 (T - Q_1 x_1 - Q_2 x_2) - d_2 Q_2 - Q_2 r_2 \left(1 - \frac{x_1 + x_2}{K}\right). \end{cases}$$
(A.9)

In case 3, the boundary equilibrium is $E_2^2 = \left(\frac{T}{q_1} - \frac{d_1}{c_1}, 0, q_1, \frac{c_2 K q_1^2 d_1}{K q_1 c_1 (d_2 + r_2) - r_2 (c_1 T - d_1 q_1)}\right)$ and the coexistence equilibrium is $E_*^2 = (x_1^{*2}, x_2^{*2}, Q_1^{*2}, Q_2^{*2})$, where

$$\begin{aligned} x_1^{*2} &= \frac{d_1 q_1 (d_2 + c_2 K) - c_1 d_2 T}{q_1 (d_1 c_2 - d_2 c_1)}, \quad x_2^{*2} &= K - \frac{d_1 q_1 (d_2 + c_2 K) - c_1 d_2 T}{q_1 (d_1 c_2 - d_2 c_1)}, \\ Q_1^{*2} &= q_1, \quad Q_2^{*2} &= \frac{d_1 q_1 c_2}{c_1 d_2}. \end{aligned}$$

The Jacobian matrix at (x_1, x_2, Q_1, Q_2) is

$$J(x_1, x_2, Q_1, Q_2) = \begin{pmatrix} r_1 \left(1 - \frac{q_1}{Q_1}\right) & 0 & \frac{r_1 x_1 q_1}{Q_1^2} & 0 \\ -\frac{r_2 x_2}{K} & r_2 \left(1 - \frac{2 x_2 + x_1}{K}\right) & 0 & 0 \\ -c_1 Q_1 & -c_1 Q_2 & -c_1 x_1 - d_1 - r_1 & -c_1 x_2 \\ \frac{Q_2 r_2}{K} - c_2 Q_1 & Q_2 \left(\frac{r_2}{K} - c_2\right) & -c_2 x_1 & \frac{r_2 (x_1 + x_2)}{K} - c_2 x_2 - d_2 - r_2 \end{pmatrix}.$$

It is verified that E_0^2 and E_1^2 do not exist.

Theorem 5. If $0 < \frac{T}{q_1} - \frac{d_1}{c_1} < K$ and $K < \frac{(c_1T - d_1q_1)(c_2q_1d_1 + c_1r_2q_2)}{c_1^2q_1q_2(d_2 + r_2)}$, E_2^2 exists and it is unstable.

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The Jacobian matrix at E_2^2 is Proof

$$J(E_2^2) = \begin{pmatrix} 0 & 0 & \frac{r_1 x_1}{q_1} & 0 \\ 0 & r_2 \left(1 - \frac{x_1}{K}\right) & 0 & 0 \\ -c_1 q_1 & -c_1 Q_2 & -c_1 x_1 - d_1 - r_1 & 0 \\ \frac{Q_2 r_2}{K} - c_2 q_1 & Q_2 \left(\frac{r_2}{K} - c_2\right) & -c_2 x_1 & \frac{r_2 x_1}{K} - d_2 - r_2 \end{pmatrix}$$

The eigenvalues of $J(E_2^2)$ are

$$\lambda_{1} = r_{2} \left(1 - \frac{\frac{T}{q_{1}} - \frac{d_{1}}{c_{1}}}{K} \right) > 0, \quad \lambda_{2} = \frac{r_{2}}{K} \left(\frac{T}{q_{1}} - \frac{d_{1}}{c_{1}} \right) - d_{2} - r_{2},$$

$$\lambda_{3} = \frac{-\left(\frac{c_{1}T}{q_{1}} + r_{1}\right) - \sqrt{\left(\frac{c_{1}T}{q_{1}} - r_{1}\right)^{2} + 4r_{1}d_{1}}}{2} < 0,$$

$$\lambda_{4} = \frac{-\left(\frac{c_{1}T}{q_{1}} + r_{1}\right) + \sqrt{\left(\frac{c_{1}T}{q_{1}} - r_{1}\right)^{2} + 4r_{1}d_{1}}}{2}.$$

Thus, $\lambda_1 > 0$ and E_2^2 is unstable. The existence and local stability of coexistence equilibrium $E_*^2 = (x_1^{*2}, x_2^{*2}, Q_1^{*2}, Q_2^{*2})$ is analyzed below, where

$$\begin{aligned} x_1^{*2} &= \frac{d_1 q_1 (d_2 + c_2 K) - c_1 d_2 T}{q_1 (d_1 c_2 - d_2 c_1)}, \quad x_2^{*2} &= K - \frac{d_1 q_1 (d_2 + c_2 K) - c_1 d_2 T}{q_1 (d_1 c_2 - d_2 c_1)}, \\ Q_1^{*2} &= q_1, \quad Q_2^{*2} &= \frac{d_1 q_1 c_2}{c_1 d_2}. \end{aligned}$$

Theorem 6. For the case of $d_1 > \frac{c_1 d_2 q_2}{c_2 q_1}$, $0 < \frac{d_1 q_1 (d_2 + c_2 K) - c_1 d_2 T}{q_1 (d_1 c_2 - d_2 c_1)} < K$, $d_1 c_2 - d_2 c_1 \neq 0$, E_*^2 exists. E_*^2 is locally asymptotically stable if $d_1 < \frac{c_1 d_2}{c_2}$.

The Jacobian matrix at E_*^2 is Proof

$$J(E_*^2) = \begin{pmatrix} 0 & 0 & \frac{r_1 x_1^{*2}}{q_1} & 0 \\ -\frac{r_2 x_2^{*2}}{K} & -\frac{r_2 x_2^{*2}}{K} & 0 & 0 \\ -c_1 q_1 & -c_1 Q_2^{*2} & -c_1 x_1^{*2} - d_1 - r_1 & -c_1 x_2^{*2} \\ \frac{Q_2^{*2} r_2}{K} - c_2 q_1 & Q_2^{*2} \left(\frac{r_2}{K} - c_2\right) & -c_2 x_1^{*2} & -c_2 x_2^{*2} - d_2 \end{pmatrix}$$

The corresponding characteristic equation is

$$f(\lambda) = \lambda^4 + B_1 \lambda^3 + B_2 \lambda^2 + B_3 \lambda + B_4,$$

where

$$B_{1} = \frac{r_{2}x_{2}^{*2}}{K} + c_{1}x_{1}^{*2} + c_{2}x_{2}^{*2} + d_{1} + d_{2} + r_{1},$$

$$B_{2} = \frac{r_{2}x_{2}^{*2}(c_{1}x_{1}^{*2} + d_{1} + c_{2}x_{2}^{*1} + d_{2} + r_{1})}{K} + c_{2}x_{2}^{*2}d_{1} + c_{2}x_{2}^{*2}r_{1} + c_{1}x_{1}^{*2}d_{2} + c_{1}r_{1}x_{1}^{*2} + d_{1}d_{2} + d_{2}r_{1},$$

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$$B_{3} = \frac{r_{2}x_{2}(c_{2}x_{2}^{*2}d_{1} + c_{2}x_{2}^{*2}r_{1} + c_{1}x_{1}^{*2}d_{2} + c_{1}r_{1}x_{1}^{*2} + d_{1}d_{2} + d_{2}r_{1})}{K} + r_{1}c_{1}d_{2}x_{1}^{*2},$$

$$B_{4} = \frac{c_{1}r_{1}r_{2}x_{1}^{*2}x_{2}^{*2}d_{2}(q_{1} - Q_{2}^{*2})}{q_{1}K}.$$

We verify that

$$B_1 > 0$$
, $B_1B_2 - B_3 > 0$, and $B_1B_2B_3 - B_1^2B_4 - B_3^2 > 0$.

If $d_1 < \frac{c_1d_2}{c_2}$, $B_4 > 0$. According to the Routh-Hurwitz criterion [24], if $d_1 < \frac{c_1d_2}{c_2}$, E_*^2 is locally asymptotically stable.

Case 4.
$$\frac{x_1 + x_2}{K} < \frac{q_1}{Q_1}, \frac{x_1 + x_2}{K} < \frac{q_2}{Q_2}$$

$$\begin{cases} \frac{dx_1}{dt} = r_1 x_1 \left(1 - \frac{q_1}{Q_1}\right), \\ \frac{dx_2}{dt} = r_2 x_2 \left(1 - \frac{q_2}{Q_2}\right), \\ \frac{dQ_1}{dt} = c_1 (T - Q_1 x_1 - Q_2 x_2) - d_1 Q_1 - Q_1 r_1 \left(1 - \frac{q_1}{Q_1}\right), \\ \frac{dQ_2}{dt} = c_2 (T - Q_1 x_1 - Q_2 x_2) - d_2 Q_2 - Q_2 r_2 \left(1 - \frac{q_2}{Q_2}\right). \end{cases}$$
(A.10)

In Case 4, (A.10) does not have coexistence equilibrium. The boundary equilibria are

$$\begin{split} E_0^3 &= \left(0, 0, \frac{c_1 T + r_1 q_1}{d_1 + r_1}, \frac{c_2 T + r_2 q_2}{d_2 + r_2}\right), \\ E_1^3 &= \left(0, \frac{T}{q_2} - \frac{d_2}{c_2}, \frac{c_1 d_2 q_2 + c_2 r_1 q_1}{c_2 (d_1 + r_1)}, q_2\right), \\ E_2^3 &= \left(\frac{T}{q_1} - \frac{d_1}{c_1}, 0, q_1, \frac{c_2 d_1 q_1 + c_1 r_2 q_2}{c_1 (d_2 + r_2)}\right). \end{split}$$

The Jacobian matrix at (x_1, x_2, Q_1, Q_2) is

$$J(x_1, x_2, Q_1, Q_2) = \begin{pmatrix} r_1 \left(1 - \frac{q_1}{Q_1}\right) & 0 & \frac{r_1 x_1 q_1}{Q_1^2} & 0 \\ 0 & r_2 \left(1 - \frac{q_2}{Q_2}\right) & 0 & \frac{r_2 x_2 q_2}{Q_2^2} \\ -c_1 Q_1 & -c_1 Q_2 & -c_1 x_1 - d_1 - r_1 & -c_1 x_2 \\ -c_2 Q_1 & -c_2 Q_2 & -c_2 x_1 & -c_2 x_2 - d_2 - r_2 \end{pmatrix}.$$

In the following analysis, let $T_1 = \frac{d_1q_1}{c_1}$ and $T_2 = \frac{d_2q_2}{c_2}$.

Theorem 7. If $T \ge max\{T_1, T_2\}$, E_0^3 exists. If $T > T_1$ or $T > T_2$, E_0^3 is unstable in $\Phi(\Omega)$. *Proof* The Jacobian matrix at E_0^3 is

$$J(E_0^3) = \begin{pmatrix} r_1 \left(1 - \frac{q_1 d_1 + q_1 r_1}{c_1 T + r_1 q_1}\right) & 0 & 0 & 0\\ 0 & r_2 \left(1 - \frac{q_2 d_2 + q_2 r_2}{c_2 T + r_2 q_2}\right) & 0 & 0\\ -\frac{c_1^2 T + c_1 r_1 q_1}{d_1 + r_1} & -\frac{c_1 c_2 T + c_1 r_2 q_2}{d_2 + r_2} & -d_1 - r_1 & 0\\ -\frac{c_1 c_2 T + c_2 r_1 q_1}{d_1 + r_1} & -\frac{c_2^2 T + c_2 r_2 q_2}{d_2 + r_2} & 0 & -d_2 - r_2 \end{pmatrix}$$

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The eigenvalues of $J(E_0^3)$ are

$$\begin{split} \lambda_1 &= r_1 \Big(1 - \frac{q_1 d_1 + q_1 r_1}{c_1 T + r_1 q_1} \Big), \quad \lambda_2 &= r_2 \Big(1 - \frac{q_2 d_2 + q_2 r_2}{c_2 T + r_2 q_2} \Big), \\ \lambda_3 &= -d_1 - r_1 < 0, \quad \lambda_4 &= -d_2 - r_2 < 0. \end{split}$$

If $T > T_1$ or $T > T_2$, $\lambda_1 > 0$ or $\lambda_2 > 0$, E_0^3 is unstable. If $T = T_1$ or $T = T_2$, the stability cannot be determined.

Theorem 8. If $T_1 \leq T_2 < T < q_2K + T_2$ and $K > \frac{(c_2T - q_2d_2)(c_1d_2q_2 + c_2r_1q_1)}{q_1q_2c_2^2(d_1+r_1)}$, E_1^3 exists. If $T_2 > T_1$, E_1^3 is unstable in $\Phi(\Omega)$.

Proof The Jacobian matrix at E_1^3 is

$$J(E_1^3) = \begin{pmatrix} r_1 \left(1 - \frac{q_1 c_2 d_1 + r_1 q_1 c_2}{c_1 d_2 q_2 + c_2 r_1 q_1}\right) & 0 & 0 & 0\\ 0 & 0 & 0 & \frac{r_2 T}{q_2^2} - \frac{d_2 r_2}{c_2 q_2} \\ - \frac{c_1^2 d_2 q_2 + c_1 c_2 r_1 q_1}{c_2 (d_1 + r_1)} & -c_1 q_2 & -d_1 - r_1 & \frac{c_1 d_2}{c_2} - \frac{c_1 T}{q_2} \\ - \frac{c_1 c_2 d_2 q_2 + c_2^2 r_1 q_1}{c_2 (d_1 + r_1)} & -c_2 q_2 & 0 & -\frac{c_2 T}{q_2} - r_2 \end{pmatrix}.$$

The eigenvalues of $J(E_1^3)$ are

$$\lambda_{1} = r_{1} \left(1 - \frac{q_{1}c_{2}d_{1} + r_{1}q_{1}c_{2}}{c_{1}d_{2}q_{2} + c_{2}r_{1}q_{1}} \right), \quad \lambda_{2} = -d_{1} - r_{1} < 0,$$

$$\lambda_{3} = \frac{-\left(\frac{c_{2}T}{q_{2}} + r_{2}\right) - \sqrt{\left(\frac{c_{2}T}{q_{2}} - r_{2}\right)^{2} + 4r_{2}d_{2}}}{2} < 0,$$

$$\lambda_{4} = \frac{-\left(\frac{c_{2}T}{q_{2}} + r_{2}\right) + \sqrt{\left(\frac{c_{2}T}{q_{2}} - r_{2}\right)^{2} + 4r_{2}d_{2}}}{2} < 0.$$

If $T_2 > T_1$ and $\lambda_1 > 0$, E_1^3 is unstable. If $T_2 = T_1$, the stability of E_1^3 cannot be determined.

Theorem 9. If $T_2 \leq T_1 < T < q_1K + T$ and $K > \frac{(c_1T - q_1d_1)(c_2d_1q_1 + c_1r_2q_2)}{q_1q_2c_1^2(d_2+r_2)}$, E_2^3 exists. If $T_1 > T_2$, E_2^3 is unstable in $\Phi(\Omega)$.

Proof The Jacobian matrix at E_2^3 is

$$J(E_2^3) = \begin{pmatrix} 0 & 0 & \frac{r_1T}{q_1^2} - \frac{d_1r_1}{c_1q_1} & 0 \\ 0 & r_2\left(1 - \frac{q_2c_1d_2 + r_2q_2c_1}{c_2d_1q_1 + c_1r_2q_2}\right) & 0 & 0 \\ -c_1q_1 & -\frac{c_1c_2d_1q_1 + c_1^2r_2q_2}{c_1d_2 + r_2)} & -\frac{c_1T}{q_1} - r_1 & 0 \\ -c_2q_1 & -\frac{c_2^2d_1q_1 + c_1c_2r_2q_2}{c_1(d_2 + r_2)} & -c_2\left(\frac{T}{q_1} - \frac{d_1}{c_1}\right) & -d_1 - r_1 \end{pmatrix}$$

The eigenvalues of $J(E_2^3)$ are

$$\lambda_1 = r_2 \left(1 - \frac{q_2 c_1 d_2 + r_2 q_2 c_1}{c_2 d_1 q_1 + c_1 r_2 q_2} \right), \quad \lambda_2 = -d_2 - r_2 < 0,$$

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$$\lambda_{3} = \frac{-\left(\frac{c_{1}T}{q_{1}} + r_{1}\right) - \sqrt{\left(\frac{c_{1}T}{q_{1}} - r_{1}\right)^{2} + 4r_{1}d_{1}}}{2} < 0,$$
$$\lambda_{4} = \frac{-\left(\frac{c_{1}T}{q_{1}} + r_{1}\right) + \sqrt{\left(\frac{c_{1}T}{q_{1}} - r_{1}\right)^{2} + 4r_{1}d_{1}}}{2} < 0.$$

If $T_1 > T_2$, $\lambda_1 > 0$, E_2^3 is unstable. If $T_1 = T_2$, the stability of E_2^3 cannot be determined.



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